

1 **ENRICHED FUNCTIONAL CARROT SMOOTHIE WITH BIOSYNTHESED**
2 **PHENOLIC COMPOUNDS**

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14

15 **Abstract**

16 Carrots are worldwide highly consumed due to their sensory characteristics and health-
17 promoting properties. However, their low phenolic/antioxidant levels may be greatly
18 increased with abiotic stresses. Accordingly, phenolic/antioxidant enhancements and
19 related phenylalanine ammonia-lyase (PAL) activity of carrot shreds under different
20 treatments (UVc radiation and hyperoxia conditions) were studied after a pre-
21 enrichment incubation of 72 h at 15°C. Subsequently, a carrot smoothie prepared from
22 enriched carrot shreds was heat-treated (90°C for 30 s) and stored at 5°C up to 14 days.
23 Heat-treated smoothies showed a good physicochemical and microbiological quality (<6
24 log CFU g⁻¹) after 14 days at 5°C, although those non-irradiated samples registered
25 lower psychrophilic and yeasts and molds levels. Heat-treated smoothie from non-

26 irradiated shreds stored under hyperoxia conditions, showed the highest total phenolic
27 content of 13,82 mg ChAE kg⁻¹ fw (87 % chlorogenic acid) after 14 days at 5°C. Total
28 phenolics content was in accordance with PAL and total antioxidant capacity.
29 Conclusively, a pre-enrichment incubation of carrot shreds under hyperoxia conditions
30 allowed to obtain a functional smoothie with great phenolic levels and good
31 microbiological and physicochemical quality up to 14 days at 5°C.

32

33 **Keywords:** wounding; hyperoxia; UVc radiation; antioxidant.

34

35 **1. INTRODUCTION**

36 The crescent consumer's knowledge on functional foods has led to an increasing interest
37 in foods not only intended to feed, but also to prevent chronic and nutritional-related
38 diseases as well as to improve overall human well-being. High intake of fruit and
39 vegetables has been proved to prevent a grand array of diseases, such as degenerative
40 disorders, cancer and cardiovascular among others (Slavin and Lloyd 2012). However,
41 the current lifestyle turns difficult the preparation of these plant products. Thus, their
42 consumption should be promoted through the development of attractive ready-to-eat
43 products that should be processed with minimal and non-aggressive treatments to
44 preserve as much as possible the quality parameters of the raw materials (Artés et al.
45 2009). Accordingly, smoothies represent an excellent and convenient alternative to
46 promote the daily consumption of fruits and vegetables (Rodríguez-Verástegui et al.
47 2015). Smoothies are nonalcoholic beverages prepared from fresh or frozen fruit and/or
48 vegetables, which are blended without filtering and usually mixed with crushed ice to
49 be immediately consumed. Often, some smoothies may include other components like
50 yogurt, milk, ice-cream, lemonade or tea. They have a milk shake-like consistency that

51 is thicker than slush drinks (Castillejo et al. 2015). Fruits and vegetables are rich in
52 phenolic compounds among other bioactive compounds. Phenolic compounds are great
53 antioxidants related to several health-promoting properties such as anti-inflammatory,
54 antitumoral, as well as preventing neurodegenerative and chronic disorders (El Gharras
55 2009). Phenylalanine ammonia-lyase (PAL) is the key enzyme of primary (shikimate)
56 and secondary (phenylpropanoid) pathways and is, therefore, involved in the
57 biosynthesis of polyphenolic compounds (Dixon and Paiva 1995). It is well reported
58 that PAL activity may be enhanced by an array of biotic and abiotic stress-induced
59 mechanisms, such as wounding, radiation exposure, hyperoxia storage, water stress,
60 ultrasounds, chilling injury, low minerals, hormones and pathogen attack, among others
61 (Cisneros-Zevallos 2003; Cuéllar-Villarreal et al. 2016). Previous studies have shown
62 that wounding, low UVc doses and hyperoxia storage, singly, enhance phenolic content
63 on carrots and other foodstuff (Alegria et al. 2012; Avena-Bustillos et al. 2012; Becerra-
64 Moreno et al. 2012; Jacobo-Velázquez et al. 2011). Carrots occupy the sixth place
65 among the list of most consumed vegetables in the American diet, although its total
66 phenolics content (TPC) is almost the lowest among them (Chun et al. 2005).
67 Accordingly, the health benefits derived from carrots could be increased by enhancing
68 their phenolics levels during a controlled pre-enrichment incubation by using
69 postharvest abiotic stresses. Furthermore, synergistic effects on phenolics increments
70 after combined application of different stresses may occur. Nevertheless, the effects on
71 phenolic/antioxidant levels after combined application of wounding, intermediate UVc
72 dose and hyperoxia atmosphere on carrots has not been studied yet. Therefore, a
73 functional phenolic-enriched carrot smoothie may be developed previously applying
74 abiotic stresses on carrot material, singled or combined, during a pre-enrichment
75 incubation prior to smoothie preparation. Moreover, a mild heat treatment of the

76 smoothie may guarantee the food safety criteria and physicochemical quality of this
77 functional carrot smoothie during refrigerated storage.

78 Accordingly, the aim of this study was to optimize a pre-enrichment treatment of carrots
79 to maximize the phenolic/antioxidant levels in order to obtain a functional carrot
80 smoothie with enhanced phenolic/antioxidants contents. Furthermore, the effects of a
81 mild heat treatment and subsequent refrigerated storage on the enriched
82 phenolic/antioxidant levels of the functional carrot smoothie were also studied.

83

84 **2. MATERIALS AND METHODS**

85 **2.1. Plant material**

86 Fresh carrots (*Daucus carota* L. cv. Nantes) were purchased at a local market in
87 Cartagena (Southeast of Spain) in April. Carrots were carefully inspected, selecting
88 those with similar visual appearance and size (14-15 cm long and 2-3 cm diameter).
89 Subsequently, carrots were sanitized in a cold room (8°C) with chlorine (100 ppm
90 NaClO; 5°C; pH 6.5±0.1) for 2 min, rinsed with tap water at 5°C for 1 min and drained
91 in a perforated basket for 1 min. A ratio of 300 g plant material: 5 L chlorine was used.
92 Carrots were wounded to shreds (2mm×3mm×40-60 mm) with a food processor
93 (FreshExpress+, Moulinex, Lyon, France). Pre-enrichment treatments were conducted
94 immediately after wounding.

95

96 **2.2. Pre-treatment treatments and incubation of plant material**

97 The UVc treatment chamber used was detailed by Artés-Hernández et al. (2009). Carrot
98 shreds were placed between the two lines of UVc lamps at 17.5 cm above and below
99 over a 35 mm thick bi-oriented PP film mounted on polystyrene net (130×68 cm) that
100 minimized blockage of the UVc radiation. The applied UVc intensity of 67.6 Wm² was

101 calculated as the mean of 18 UVc readings on each side of the net using a VLX 254
102 radiometer (Vilber Lourmat, Marne la Vallee, France). Thus, both sides received the
103 same UVc intensity. The UVc light intensity was kept constant and the applied dose
104 was varied by altering the exposure time at the fixed distance. A UVc radiation
105 treatment of 4 kJ UVc m⁻² (exposure time of 139 s) was applied. Non-irradiated samples
106 were used as control (hereinafter 'CTRL').

107 Samples to be stored under hyperoxia conditions (hereinafter 'HO') were placed in
108 plastic containers (30 cm diameter, 60 cm height) and connected to an air-flow-through
109 system supplied with humidified flows of either air or a gas mixture containing 80 % O₂
110 (balanced with N₂). In order to ensure a good air flow through carrot shreds, these
111 samples were distributed in opened plastic petri dishes (8.5 cm diameter, 1 cm height).
112 CO₂ partial pressures were kept <0.15 kPa to avoid any physiological effect exerted by
113 CO₂ such as anaerobic metabolism. Samples stored under air conditions were used as
114 control (hereinafter 'Air'). Gas treatments were applied at 15°C for 72 h in darkness.
115 Pre-enrichment incubation of carrots, as well as smoothie preparation and subsequent
116 storage, is summarized in Figure 1.

117

118 **2.3. Carrot smoothie preparation, heat treatment and storage conditions**

119 Carrot smoothie was prepared in a food processor (Robot Cook[®], Robot Coupe,
120 Vincennes Cedex, France) using sterilized water in a relation of 1:1 (carrot weight:
121 water volume). Heat treatment of carrot smoothie was applied using a Mastia
122 thermoresistometer described by Conesa et al. (2009). Immediately after smoothie
123 blending, the sterilized vessel of the thermoresistometer was filled with 400 mL of the
124 smoothie. The thermoresistometer was programmed to increase the initial smoothie
125 temperature (8±2°C) with a heating rate of 30°C/min to 90°C, then maintained for 30 s

126 and cooled down to a final temperature of 40°C (cooling rate of 30°C/min). After heat
127 treatment, the smoothie temperature was cooled down to 5°C submerging the vessel in
128 an ice-water bath while continuously agitation was programmed in the
129 thermoresistometer. Subsequently, 15-mL aliquots of heat-treated samples were taken in
130 aseptic conditions in sterile Falcon tubes through the thermoresistometer sampling port.
131 Samples were stored in darkness at 5°C. Non heat-treated carrot smoothie was used as
132 control. Visual appearance, flavor, texture, off-colors, off-odors, lumpiness, turbidity,
133 precipitation/phase separation and overall quality of heat-treated smoothie conducted by
134 an informal sensory panel test of 8 persons were reported to be over the limit of
135 acceptability up to 14 days at 5 °C. Sampling was conducted on processing day (0) and
136 after 7 and 14 days at 5°C. Five replicates per pre-treatment and sampling day were
137 prepared.

138

139 **2.4. Analyses**

140 2.4.1. Physiochemical analyses

141 The pH, titratable acidity (TA), total soluble solids content (SSC) and color of samples
142 were determined according to Castillejo et al. (2015). TA and SSC were expressed as g
143 citric acid 100 mL⁻¹ and °Brix, respectively. Total color differences (ΔE) and browning
144 index differences (ΔBI) were calculated according to equations previously described
145 (Palou et al. 1999).

146

147 2.4.2. Microbial analysis

148 Mesophilic, psychrophilic and yeast and mold growth was determined using standard
149 enumeration methods according to Castillejo et al. (2015). All microbial counts were
150 reported as log colony forming units per gram of product (log CFU g⁻¹). Each of the

151 five replicates was analyzed by duplicate. The presence of *Salmonella* spp., *Listeria*
152 *monocytogenes* and generic *Escherichia coli* was monitored throughout storage of
153 smoothies according to the European legislation (EC_1441/2007 2007).

154

155 2.4.3. Phenylalanine ammonia-lyase

156 PAL activity was analyzed according to Ke and Saltveit (1986) with modifications
157 (Jacobo-Velázquez et al. 2011). Concisely, 2 g of sample was mixed with
158 polyvinylpyrrolidone (0.2 g) and homogenized in cold 50 mM borate buffer (pH
159 8.5) containing 400 $\mu\text{L L}^{-1}$ β -mercaptoethanol. Homogenates were filtered through four
160 layers of cheesecloth and then centrifuged at 10,000 \times G for 20 min at 4°C. Supernatants
161 were used as enzyme extract. Two sets of UV-Star well plates (Greiner Bio-One,
162 Frickenhausen, Germany) containing 69 μL of PAL extract plus 200 μL ultrapure water
163 were prepared for every sample and pre-incubated at 40 °C for 5 min. Afterwards, 30 μL
164 of either water (blank) or 100 mM L-phenylalanine substrate solution (freshly prepared
165 before assay) were added to each of the well for every sample set. The absorbances of
166 sample sets were measured at 290 nm using a Multiscan plate reader (Tecan Infinite
167 M200, Männedorf, Switzerland) at time 0 and after 1 h of incubation at 40 °C. The PAL
168 activity was expressed as μmol of *t*-cinnamic acid synthesized kg^{-1} fw h^{-1} using a *t*-
169 cinnamic acid standard curve (0-6.75 mM). Each of the three replicates was analyzed by
170 duplicate.

171

172 2.4.4. Phenolic compounds

173 Extraction to determine phenolic compounds and total antioxidant capacity (TAC)
174 extract was conducted by homogenization (Ultra Turrax[®] model 18T, IKA-Werke
175 GmbH & Co. KG, Germany) of 2 g of sample in 8 mL methanol for 20 s under ice-

176 water bath. Subsequently, extracts were centrifuged at 13500×G for 20 min at 4 °C and
177 supernatants were collected and analyzed. Extracts for individual phenolic compounds
178 were further filtered through a 0.22 µm polyethersulphone filter and stored at –80 °C in
179 amber vials until UPLC analysis.

180 Total phenolic content (TPC) was analyzed by Folin–Ciocalteu reagent method as
181 previously described (Martínez-Hernández et al. 2011). Briefly, a 19 µL aliquot of TPC
182 extract was placed on a 96 PS flat bottom well plate (Greiner Bio-One, Frickenhausen,
183 Germany) and 29 µL of Folin–Ciocalteu reagent 2 N (Sigma, St Louis, MO, USA) were
184 added. Samples were incubated for 3 min in darkness at room temperature. After
185 incubation, 192 µL of a solution containing Na₂CO₃ (4 g L⁻¹) and NaOH (20 g L⁻¹) were
186 added and the reaction was carried out for 1 h at room temperature in darkness.
187 Subsequently, absorbance was read at 750 nm using the same microplate reader as
188 described before. TPC was expressed as chlorogenic acid (Sigma, St Louis, MO, USA)
189 equivalents (ChAE) in mg kg⁻¹ fresh weight (fw). Each of the three replicates was
190 analyze by duplicate.

191 Analyses of individual phenolic compounds were conducted as previously described
192 (Alegria 2015) with some modifications. Briefly, samples of 20 µL were analyzed using
193 an Ultra High-Performance liquid chromatography (UPLC) instrument (Shimadzu,
194 Kyoto, Japan) equipped with a DGU-20A degasser, LC-30AD quaternary pump, SIL-
195 30AC autosampler, CTO-10AS column heater and SPDM-20A photodiode array
196 detector. The UPLC system was controlled by the software LabSolutions (Shimadzu, v.
197 5.42 SP5). Chromatographic analyses were carried out onto a Kinetex C18 column (100
198 mm×4.6 mm, 2.6 µm particle size; Phenomenex, Macclesfield, UK) with a KrudKatcher
199 Ultra HPLC guard column (Phenomenex, Macclesfield, UK). The column temperature
200 was maintained at 25 °C. The mobile phase was acidified water (A; formic acid to final

201 pH 2.3) and acidified methanol (B; formic acid to final pH 2.3). The flow rate was 1.5
202 mL min⁻¹. Gradient program used was 0/88, 1.2/88, 2.4/85, 8.3/70, 9.4/50, 11.8/50,
203 20.8/55, 22.0/60 (min/% phase A). Then, column equilibration was conducted at 0 % A
204 for 2.2 min. Chromatograms were recorded at 320 nm. Phenolic acids were quantified
205 as standards of chlorogenic acid (3-CQA), ferulic acid (Sigma, St Louis, MO, USA),
206 isochlorogenic acid A (3,5-CQA) and C (4,5-CQA) (ChromaDex, Irvine, CA, USA).
207 The calibration curves were made with at least six data points. The results were
208 expressed as mg kg⁻¹ fw. Each of the three replicates was analyzed by duplicate.

209

210 2.4.5. Total antioxidant capacity

211 The extracts were analyzed for TAC according to Brand-Williams et al. (1995) with
212 modifications (Martínez-Hernández et al. 2013). Briefly, a solution of 0.7 mM 2,2-
213 diphenyl-1-picrylhydrazil (DPPH) in methanol was prepared 2 h before the assay and
214 adjusted to 1.10±0.02 nm immediately before use. A 21 µL aliquot of the previously
215 described extract was placed on a 96 PS flat-bottom well plate and 194 µL of DPPH
216 was added. The reaction was carried out for 30 min at room temperature in darkness and
217 the absorbance at 515 nm was measured using the Multiscan plate reader. Results were
218 expressed as mg Trolox equivalent antioxidant capacity kg⁻¹ fw. Each of the three
219 replicates was analyze by duplicate.

220

221 **2.5. Statistical Analyses**

222 A complete randomized design in triplicate, with two-way ANOVA
223 (treatment×storage), by Post Hoc Tuckey HSD tests, were used with SPSS software (v.
224 21, IBM, USA).

225

226 **3. RESULTS**

227 **3.1. Physicochemical quality**

228 Carrot shreds showed an initial pH and TA of 6.32-6.37 and 0.32-0.48 % (Table 1),
229 respectively, similar to previous data (Pushkala et al. 2012). The pH of the shreds
230 decreased and TA increased during pre-enrichment incubation as a combined effect of
231 microbial growth and phenolic acids enrichment as shown later. Accordingly, pH/TA of
232 non-irradiated and irradiated shreds decreased to 3.20-4.05/4.75-5.22 and 4.87-
233 5.02/3.42-4.02, respectively, after pre-enrichment incubation. The higher acidification
234 observed in non-irradiated samples may be explained by the higher phenolic acids
235 content of these samples as it will be discussed later. No clear influence of the
236 atmosphere storage conditions on pH and TA of carrot shreds after pre-enrichment
237 incubation was observed. Similarly, no significant differences on pH and TA values
238 were observed among air or hyperoxia-stored (80 % O₂) blueberry fruit during storage
239 up to 35 days at 4 °C (Zheng et al. 2003). Carrot smoothies from CTRL-Air, CTRL-HO,
240 UVc-Air and UVc-HO carrot shreds showed initial pH/TA values of 3.35/2.05,
241 4.13/1.07, 5.03/0.68 and 5.15/1.00, respectively. Heat treatment did not change ($p<0.05$)
242 initial pH and TA of carrot smoothies. Quality of carrot beverages is difficult to
243 maintain during storage due to its low acidity. The pH of carrot beverages is usually
244 acidified with citric acid, or other acidulants, to approximately 3.8 as a general
245 commercial practice by the food industries in order to reduce microbial growth and
246 degradative enzymatic and non-enzymatic reactions during storage (Quitão-Teixeira et
247 al. 2009; Talcott and Howard 1999). Alternatively, acidification of carrot juice through
248 fermentation has been proposed as a preservation method combined with pasteurization
249 (Tamminen et al. 2013). Accordingly, the spontaneous fermentation occurred during
250 pre-enrichment incubation of carrot shreds allowed to naturally reduce the pH extending

251 the shelf-life with a desirable mild acidic taste. In general, heat-treated carrot smoothies
252 did not show great pH/TA changes throughout storage at 5°C with final pH/TA values
253 of 3.8-4.5/0.87-1.14. Similarly, acidified blanched carrot juice showed more stable pH
254 and TA values than non-acidified juices up to 21 days of storage at 4°C (Yu and
255 Rupasinghe 2012).

256 Carrot shreds showed initial SSC of 7.85-7.92 (Table 1) similar to previous data
257 (Martínez-Hernández et al. 2016). Carrot shreds stored under hyperoxia conditions
258 presented higher SSC compared to air-stored samples after pre-enrichment incubation.
259 Similarly, SSC of blueberry fruit increased during hyperoxia (80 % O₂) storage at 5°C
260 (Zheng et al. 2003). The latter behavior may be explained by the reduced microbial
261 growth under hyperoxia conditions, as shown later, and consequently lower microbial
262 sugars consumption. SSC of carrot smoothies was not greatly changed after heat
263 treatment. In general, SSC of non-heat-treated carrot smoothies decreased through
264 storage as consequence of microbial growth. Contrary, SSC of heat-treated smoothies
265 generally did not register great changes due to the lower microbial loads. Kaur and
266 Sharma (2013) also reported unchanged SSC in pasteurized carrot juice after 15 days at
267 5 °C.

268 Color is an important parameter for conformity determination of carrot beverages
269 quality. ΔE is a colorimetric parameter extensively used to characterize the variation of
270 colors during processing and storage of food products. BI represents the purity of brown
271 color and is reported as an important parameter in processes where enzymatic or non-
272 enzymatic browning take place (Palou et al. 1999). For that reason, ΔE and BI have
273 been satisfactorily used to assess color quality of carrot beverages after processing
274 treatments and subsequent storage (Kaur and Sharma 2013). UVc pre-treatment of
275 carrot shreds induced initial mild browning ($\Delta E=5.6$, $\Delta BI= 89.4$; data not shown).

276 Browning observed in some fruits and vegetables after UVc radiation has been
277 attributed to the increased peroxidase (POD) activity (Tomás-Barberán and Espín
278 2001). However, such browning of carrots shreds after the low UVc dose used was not
279 visually observed. Irradiated shreds showed higher BI after pre-enrichment incubation
280 which may be owed to the pre-activated POD during UVc pre-treatment (Table 1).
281 Furthermore, pre-enrichment incubation under hyperoxia conditions induced slightly
282 higher ΔE and ΔBI compared to carrot shreds incubated under air conditions. β -
283 carotene, the main pigment responsible of the bright orange color of carrots, is very
284 susceptible to isomerization and oxidation (Knockaert et al. 2012). Furthermore, POD
285 activity may increase under hyperoxia storage as previously reported (Yang et al. 2009).
286 Accordingly, the observed greater color degradation under hyperoxia compared to air
287 conditions may be explained by a β -carotene degradation and incremented POD
288 activity. Heat treatment of carrot smoothies induced low color changes ($\Delta E < 25$,
289 $\Delta BI < 96$) which correspond to undetected visual color changes by a trained panel test
290 (Kaur and Sharma 2013). Accordingly, β -carotene degradation in carrot puree and juice
291 was very low, or even enhanced, due to higher extractability after such heat treatment as
292 previously modeled (Lemmens et al. 2010; Marx et al. 2003; Quitão-Teixeira et al.
293 2009). Color changes of heat-treated carrot smoothies during storage were lower
294 compared to untreated smoothies which may be owed to heat inactivation of color-
295 degradative enzymes and reduced β -carotene degradation under such low storage
296 temperature. Accordingly, only 5.5 % residual POD activity was reported in carrot juice
297 after a similar heat treatment and it was even reduced to 2 % after 14 days at 4°C
298 (Quitão-Teixeira et al. 2009). Attending to pre-enrichment treatments, all smoothies
299 from irradiated carrots shreds showed slightly higher color changes after 14 days of
300 storage at 5°C. Nevertheless, all heat-treated smoothies from stressed carrot shreds

301 (CTRL-HO, UVc-Air and UVc-HO) presented a good physicochemical quality after 14
302 days of storage at 5°C.

303

304 **3.2. Microbiological quality**

305 Carrot shreds showed initial mesophilic, psychrophilic and Y+M loads of 5.4, 5.1 and
306 4.8 log CFU g⁻¹, respectively (Table 2). UVc pre-treatment reduced initial microbial
307 loads of carrots shreds by 1.1-1.3 log units. Similar microbial reductions have been
308 reported in Bimi[®] broccoli after a UVc dose of 4.5 kJ m⁻² (Martínez-Hernández et al.
309 2011). The observed sanitation effect of UVc is due to the capacity of this non-ionizing
310 radiation to induction the formation of pyrimidine dimers which distort the DNA helix
311 and block microbial cell replication. Consequently, the cells become unable to repair
312 their radiation-damaged DNA and die (Bintsis et al. 2000). Pre-enrichment incubation
313 of shreds under air conditions led to mesophilic/psychrophilic and Y+M growth of
314 3.2/2.8 and 4.7 log units, respectively, after 72 h. However, pre-enrichment incubation
315 of non-irradiated shreds under HO conditions greatly limited mesophilic/psychrophilic
316 and Y+M growth to only 1.1/1.0 and 2.7 log units, respectively, after 72 h. Likewise,
317 total viable counts were better controlled under hyperoxia (90 %) compared to air
318 storage (Amanatidou et al. 2000). The observed microbicidal effects during hyperoxia
319 storage may be explained by several factors such as the unfavorable effects on the
320 oxidation–reduction potential of the system, the oxidation of enzymes having sulfhydryl
321 groups or disulfide bridges, and the accumulation of injurious reactive O₂ species
322 (Kader and Ben-Yehoshua 2000). The sanitizing effects of UVc radiation and hyperoxia
323 storage in other fresh-cut fruit and vegetables have been previously reviewed (Artés et
324 al. 2009). UVc irradiated shreds showed greater microbial growth compared to non-
325 irradiated samples during pre-enrichment incubation. Latter detrimental effect of UVc

326 pre-treatment during storage, contrary to benefit from initial sanitation, may be
327 explained by several hypothesis: 1) repair systems such as UV-induced enzymatic
328 photorepair and expression of excision-repair genes that may restore DNA integrity in
329 exposed microbial cells (Bintsis et al. 2000). Accordingly, those microorganisms with
330 restored genetic material may show greater growth rates. 2) Pant cell disruption caused
331 by UVc radiation leads to leakage of electrolytes (Martínez-Hernández et al. 2013) such
332 as sugars which favors microbial growth.

333 In general, heat treatment reduced initial microbial loads of carrot smoothies (7-8 log
334 units) below detection limits (1 log CFU mL⁻¹ for mesophilic/psychrophilic and 2 log
335 CFU mL⁻¹ for Y+M). Accordingly, the applied heat treatment was enough to achieve
336 pasteurization levels. Microbial loads of untreated smoothies were over 10 log CFU mL⁻¹
337 after 7 days at 5 °C (data not shown). Mesophilic counts of heat-treated carrot
338 smoothies increased during storage registering final loads of 3.5-4.1 log CFU mL⁻¹,
339 without significant differences among pre-treatments, after 14 days at 5°C. Smoothies
340 from UVc-HO and UVc-Air pre-treated shreds showed the highest psychrophilic
341 growths with 4.7 and 3.5 log units increments, respectively, after 14 days at 5 °C.
342 Meanwhile, smoothies from CTRL-HO and CTRL-Air shreds registered psychrophiles
343 increments of 2.9 and 1.4 log units, respectively, after 14 days at 5 °C. As observed,
344 psychrophilic growth in pasteurized smoothies was higher as the stress level from pre-
345 enrichment incubation augmented following this order: HO>UVc>UVc+HO. Latter
346 behavior may be explained since as the stress level increased surviving microorganisms
347 after heat treatment acquired greater adaptation to grow under unfavorable conditions
348 such as low temperature storage. Similarly, heat-treated smoothies from UVc-HO and
349 UVc-Air shreds registered 0.8 and 1.3 log CFU mL⁻¹ increments, respectively, while the
350 other two pre-treatments did not register significant (p<0.05) changes after 14 days at 5

351 °C. *Salmonella* spp., *Listeria monocytogenes* and generic *Escherichia coli* were
352 monitored throughout storage of smoothies meeting European legislation limits
353 (Comission Regulation (EC) No 1441/2007 2007). Phenolic acids are known to exhibit
354 antimicrobial activity against a variety of microorganisms (Wen et al. 2003). In the
355 same line, carrot juice have shown great antilisterial properties (Beuchat and Brackett
356 1990). Application of combined preservative factors (called hurdles) is used by food
357 industries according to the hurdle technology to achieve effective preservation of foods
358 (Leistner 2000). Consequently, the good microbiological quality (microbial loads < 6 log
359 units) of all heat-treated carrot smoothies after 14 days of storage may be owed to the
360 combination of achieved acidic pH, enhanced antimicrobial compounds (phenolic acids)
361 and low storage temperature.

362

363 **3.3. Phenylalanine ammonia-lyase activity**

364 PAL is the key enzyme between primary (shikimate pathway) and secondary
365 (phenylpropanoid) metabolism pathways involved in the biosynthesis of polyphenolic
366 compounds (Dixon and Paiva 1995). Carrot shreds showed initial PAL activity of 12.5-
367 16.2 $\mu\text{mol cinnamic acid formed kg}^{-1} \text{ h}^{-1} \text{ fw}$ (Table 3) similar to previous data
368 (Martínez-Hernández et al. 2016). UVc pre-treatment did not induce significant
369 ($p < 0.05$) changes in the PAL activity of carrot shreds at day 0. PAL activity of shredded
370 carrots greatly increased after pre-enrichment period. CTRL-HO carrot shreds showed
371 the highest increments with a PAL activity of 224.9 $\mu\text{mol cinnamic acid formed kg}^{-1} \text{ h}^{-1}$
372 fw after 72 h of pre-enrichment period. The rest of pre-treatments showed PAL
373 activities ranging from 86.2 to 102.7 after pre-enrichment period without significant
374 differences among them. PAL activation after wounding and hyperoxia storage has been
375 reported as an abiotic stress response being proposed ATP and reactive oxygen species

376 as signaling molecules (Jacobo-Velázquez et al. 2011). Fresh carrot smoothie from
377 CTRL-HO shreds showed an initial PAL activity of 112.43 μmol cinnamic acid formed
378 $\text{kg}^{-1} \text{h}^{-1}$ fw while the activity of this enzyme ranged from 40.6 to 52.5 μmol cinnamic
379 acid formed $\text{kg}^{-1} \text{h}^{-1}$ fw or the rest of smoothies. Pasteurization of carrot smoothie
380 greatly reduced PAL activity by 81-95 % without significant differences ($p < 0.05$)
381 among pre-treatments. Likewise, heat treatment (70°C for 3 min) of vegetables red
382 smoothies (pH 4.4) led to reductions of PAL activities of 65-70 % (Rodríguez-
383 Verástegui et al. 2015). In general, PAL activity of smoothies decreased throughout
384 storage registering final activities of 22.4/11.7 μmol cinnamic acid formed $\text{kg}^{-1} \text{h}^{-1}$ fw
385 for smoothies from CTRL-HO shreds while the rest of samples ranged among 1.0-5.7
386 μmol cinnamic acid formed $\text{kg}^{-1} \text{h}^{-1}$ fw. PAL activation due to wounding stress occurred
387 during smoothie preparation may be greatly reduced at low storage temperatures.
388 Accordingly, PAL activity of red vegetables smoothies greatly incremented after 20
389 days at 5°C (Rodríguez-Verástegui et al. 2015). Accordingly, no PAL activation was
390 observed in the carrot smoothies in this storage period of 14 days at 5°C . Thus, the
391 decrease of PAL activity observed in the carrot smoothies throughout storage may be
392 owed to the low storage temperature and acidic pH conditions as previously reported in
393 PAL preparations (Gareth Rees and Hugh Jones 1996).

394

395 **3.4. Phenolic compounds**

396 Carrot shreds reported initial TPC of 187.3 mg CHA kg^{-1} fw (Table 3). Similar TPC
397 have been reported for the same carrot cultivar (Alegria et al. 2010). The major
398 individual phenolic compounds identified were 3-CQA, 3,5-CQA, 4,5-CQA and ferulic
399 acid (Table 4). The phenolic contents of carrots were unchanged ($p < 0.05$) immediately
400 after UVc pre-treatment. The TPC of carrot shreds increased by approximately 2060,

401 1510, 1170 and 760 % in CTRL-HO, CTR-Air, UVc-Air and UVc-HO samples,
402 respectively, after 72 h of pre-enrichment incubation. Postharvest abiotic stresses such
403 as wounding, UVc radiation and hyperoxia storage have been reported to greatly
404 increment the contents of phenolic compounds in carrots during subsequent storage
405 (Alegria et al. 2012; Martínez-Hernández et al. 2011). This phenolic biosynthesis has
406 been reported to be a consequence of PAL activation after these abiotic stresses being
407 proposed ATP and reactive oxygen species as signaling molecules (Jacobo-Velázquez et
408 al. 2011). UVc-HO showed the lowest phenolic accumulation during pre-enrichment
409 incubation among the rest of treatments probably owed to a partial PAL denaturation by
410 such UVc treatment delaying the stress-enhanced activity of this enzyme. The pre-
411 enrichment incubation of carrot shreds allowed to obtain carrot smoothies with TPC of
412 710.4-1925.7 mg CHA kg⁻¹ fw, representing 3-CQA the 87.3 % of the sum of phenolic
413 compounds. 3-CQA, an ester of caffeic acid with quinic acid with great antioxidant
414 capacity compared to other phenolic compounds, has been reported as the main
415 phenolic compound in carrots (Castelluccio et al. 1995). The identified minor phenolic
416 compounds 3,5-CQA, 4,5-CQA and ferulic acid accounted 7.8, 2.4 and 2.5 % of the
417 sum of phenolic compounds, respectively. Heat treatment of carrot smoothies did not
418 induce significant ($p < 0.05$) changes of TPC or individual phenolic compounds.
419 Consistently, no TPC changes were reported between untreated and heat-treated carrot
420 purees and juices (Patras et al. 2009; Quitão-Teixeira et al. 2009).

421 The TPC of untreated smoothies registered a mild TPC increment of 10-25 % at day 7
422 showing the smoothie from CTRL-HO shreds the highest increment. This TPC
423 increment at day 7 is in accordance to the ferulic acid and 3,5-CQA increments (Table
424 4) and to the greater PAL activity observed of these samples regarding the rest of
425 smoothies (Table 3). However, heat-treated smoothies did not show the same behavior

426 at day 7. Similar phenolic increments have been reported in red vegetables smoothies
427 during low temperature storage (Rodríguez-Verástegui et al. 2015). Interestingly, great
428 TPC increments of approximately 610-850 % were registered in heat-treated smoothies
429 at day 14 comparing to their respective initial levels. Heat-treated smoothies from non-
430 irradiated air/HO shreds showed the highest TPC at day 14 with approximately
431 10960/13824 mg CHA kg⁻¹ fw. This great TPC enhancement of non-irradiated samples
432 were due to 3-CQA and ferulic acids enhancements. The greater phenolic biosynthesis
433 observed in smoothies from air-incubated carrot shreds is in accordance to the still
434 higher PAL activities of these smoothies at day 14. However, PAL activity of those
435 samples at day 14 may not explain such great increments of phenolic compounds
436 observed in heat-treated smoothies. Accordingly, this enhanced biosynthesis of phenolic
437 compounds in heat-treated smoothies at day 14 may be owed to other enzymes different
438 to PAL involved in the phenylpropanoid pathway. Heat treatment (100 °C for 45 s) of
439 carrots has been reported to induce TPC enhancements during subsequent storage of
440 carrot shreds at 5 °C comparing to untreated samples (Alegria et al. 2012). Accordingly,
441 the heat treatment applied to the carrot smoothies could trigger signals related to other
442 enzymes different from PAL involved in the phenylpropanoid pathway although the
443 activation of these enzymes could be retarded until day 14 due to the low storage
444 temperature.

445

446 **3.5. Total antioxidant capacity**

447 The initial TAC of carrot shreds was 1102.3±97.3 mg Trolox equivalent kg⁻¹ fw (Table
448 3). TAC increased during pre-enrichment incubation being highly correlated (R²=0.90)
449 to TPC. Among pre-enrichment treatments, non-irradiated carrot shreds stored under
450 hyperoxia conditions registered the highest TAC enhancements as observed for TPC.

451 Similar high TAC-TPC correlations have been previously reported after wounding and
452 hyperoxia storage of carrots (Jacobo-Velázquez et al. 2011). The specific antioxidant
453 capacity (ratio of total antioxidant capacity over total soluble phenolics) has been
454 proposed as a useful index to provide information of the effectiveness of phenolic
455 compounds to neutralize free radicals (Cisneros-Zevallos 2003; Heredia and Cisneros-
456 Zevallos 2009). A higher specific antioxidant capacity means phenolic compounds have
457 a higher capacity to stabilize free radicals (Reyes et al. 2007). Irradiated samples
458 showed higher specific antioxidant capacities compared to non-irradiated samples
459 reporting UVc-HO shreds the highest value with 1010.7 $\mu\text{mol Trolox mg}^{-1}$ ChAE.
460 Similar specific antioxidant activity was reported by induced carrot phenolics after
461 postharvest abiotic stresses (Cisneros-Zevallos 2003).

462 Smoothies from carrots without UVc pre-treatment showed the highest TAC levels with
463 1462.2 (hyperoxia) and 1343.8 (air) $\mu\text{mol Trolox mg}^{-1}$ ChAE. On the other side,
464 smoothies from irradiated carrots showed lower TAC levels with 829.1 (hyperoxia) and
465 1056.9 (air) $\mu\text{mol Trolox mg}^{-1}$ ChAE. Similar to TPC, heat treatment of carrot
466 smoothies did not induce significant ($p < 0.05$) changes of TAC. TAC of smoothies
467 increased by 64-227 % after 14 days of storage at 5 °C. The greatest TAC increments
468 after 14 days in un-heated smoothies was observed in those samples from non-irradiated
469 carrots. However, the highest TAC increments in heat-treated smoothies were registered
470 by UVc pretreated samples. Latter behavior may be explained since UVc pre-treatment
471 was able to compensate subsequent reduction of activities of enzymes involved in the
472 phenylpropanoid pathway after heat treatment.

473

474 **5. Conclusions**

475 Carrot is a vegetable highly consumed which low phenolic levels could be naturally
476 increased leading to a phenolic/antioxidant enriched plant material to produce a
477 functional carrot smoothie. The phenolic levels of shredded carrots used for the
478 smoothie preparation were greatly enhanced after pre-enrichment incubation for 72 h at
479 15 °C up to 2060 % in those non-irradiated shreds stored under hyperoxia conditions.
480 The total antioxidant capacity was highly correlated to total phenolic content. The high
481 temperature-short time heat treatment reduced microbial loads below the detection
482 limits with low growth during subsequent refrigerated storage. The physicochemical
483 quality was good for all smoothies at the end of storage. UV-C pretreatment of carrot
484 shreds resulted in carrot smoothies at the end of storage with higher psychrophilic and
485 yeasts and molds loads and lower phenolic levels. Accordingly, pre-enrichment
486 incubation of carrot shreds under hyperoxia conditions allowed to obtain a functional
487 smoothie with great phenolic levels and good microbiological and physicochemical
488 quality up 14 days at 5 °C.

489

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508 **References**

- 509 Alegria, C. (2015). Heat shock and uv-c abiotic stress treatments as alternative tools to
510 promote fresh-cut carrot quality and shelf-life. *PhD Thesis. University of Lisbon*.
- 511 Alegria, C., Pinheiro, J., Duthoit, M., Gonçalves, E. M., Moldão-Martins, M., & Abreu, M. (2012).
512 Fresh-cut carrot (cv. Nantes) quality as affected by abiotic stress (heat shock and UV-C
513 irradiation) pre-treatments. *LWT - Food Science and Technology*, 48(2), 197-203,
514 doi:<http://dx.doi.org/10.1016/j.lwt.2012.03.013>.
- 515 Alegria, C., Pinheiro, J., Gonçalves, E. M., Fernandes, I., Moldão, M., & Abreu, M. (2010).
516 Evaluation of a pre-cut heat treatment as an alternative to chlorine in minimally
517 processed shredded carrot. *Innovative Food Science & Emerging Technologies*, 11(1),
518 155-161, doi:<http://dx.doi.org/10.1016/j.ifset.2009.10.008>.
- 519 Amanatidou, A., Slump, R. A., Gorris, L. G. M., & Smid, E. J. (2000). High Oxygen and High
520 Carbon Dioxide Modified Atmospheres for Shelf-life Extension of Minimally Processed
521 Carrots. *Journal of Food Science*, 65(1), 61-66, doi:10.1111/j.1365-
522 2621.2000.tb15956.x.
- 523 Artés-Hernández, F., Escalona, V. H., Robles, P. A., Martínez-Hernández, G. B., & Artés, F.
524 (2009). Effect of UV-C radiation on quality of minimally processed spinach leaves.
525 *Journal of the Science of Food and Agriculture*, 89(3), 414-421, doi:10.1002/jsfa.3460.
- 526 Artés, F., Gómez, P., Aguayo, E., Escalona, V., & Artés-Hernández, F. (2009). Sustainable
527 sanitation techniques for keeping quality and safety of fresh-cut plant commodities.
528 *Postharvest Biology and Technology*, 51(3), 287-296,
529 doi:<http://dx.doi.org/10.1016/j.postharvbio.2008.10.003>.
- 530 Avena-Bustillos, R. J., Du, W. X., Woods, R., Olson, D., Breksa, A. P., 3rd, & McHugh, T. H.
531 (2012). Ultraviolet-B light treatment increases antioxidant capacity of carrot products.
532 *Journal of the Science of Food and Agriculture*, 92(11), 2341-2348,
533 doi:10.1002/jsfa.5635.
- 534 Becerra-Moreno, A., Benavides, J., Cisneros-Zevallos, L., & Jacobo-Velázquez, D. A. (2012).
535 Plants as Biofactories: Glyphosate-Induced Production of Shikimic Acid and Phenolic
536 Antioxidants in Wounded Carrot Tissue. *Journal of Agricultural and Food Chemistry*,
537 60(45), 11378-11386, doi:10.1021/jf303252v.
- 538 Beuchat, L. R., & Brackett, R. E. (1990). Inhibitory effects of raw carrots on *Listeria*
539 *monocytogenes*. *Applied and Environmental Microbiology*, 56(6), 1734-1742.
- 540 Bintsis, T., Litopoulou-Tzanetaki, E., & Robinson, R. K. (2000). Existing and potential
541 applications of ultraviolet light in the food industry – a critical review. *Journal of the*

542 *Science of Food and Agriculture*, 80(6), 637-645, doi:10.1002/(SICI)1097-
543 0010(20000501)80:6<637::AID-JSFA603>3.0.CO;2-1.

544 Brand-Williams, W., Cuvelier, M., & Berset, C. (1995). Use of a free radical method to evaluate
545 antioxidant activity. *LWT-Food Science and Technology*, 28(1), 25-30.

546 Castelluccio, C., Paganga, G., Melikian, N., Bolwell, G. P., Pridham, J., Sampson, J., et al. (1995).
547 Antioxidant potential of intermediates in phenylpropanoid metabolism in higher
548 plants. *FEBS Letters*, 368(1), 188-192.

549 Castillejo, N., Martínez-Hernández, G. B., Gómez, P. A., Artés, F., & Artés-Hernández, F. (2015).
550 Red fresh vegetables smoothies with extended shelf life as an innovative source of
551 health-promoting compounds. [journal article]. *Journal of Food Science and
552 Technology*, 1-12, doi:10.1007/s13197-015-2143-2.

553 Cisneros-Zevallos, L. (2003). The Use of Controlled Postharvest Abiotic Stresses as a Tool for
554 Enhancing the Nutraceutical Content and Adding-Value of Fresh Fruits and Vegetables.
555 *Journal of Food Science*, 68(5), 1560-1565, doi:10.1111/j.1365-2621.2003.tb12291.x.

556 Comission Regulation (EC) No 1441/2007 (2007). *Official Journal of the European Union*, 322,
557 12-29.

558 Conesa, R., Andreu, S., Fernandez, P. S., Esnoz, A., & Palop, A. (2009). Nonisothermal heat
559 resistance determinations with the thermoresistometer Mastia. *Journal of Applied
560 Microbiology*, 107(2), 506-513, doi:10.1111/j.1365-2672.2009.04236.x.

561 Cuéllar-Villarreal, M. d. R., Ortega-Hernández, E., Becerra-Moreno, A., Welte-Chanes, J.,
562 Cisneros-Zevallos, L., & Jacobo-Velázquez, D. A. (2016). Effects of ultrasound treatment
563 and storage time on the extractability and biosynthesis of nutraceuticals in carrot
564 (*Daucus carota*). *Postharvest Biology and Technology*, 119, 18-26,
565 doi:<http://dx.doi.org/10.1016/j.postharvbio.2016.04.013>.

566 Chun, O. K., Kim, D.-O., Smith, N., Schroeder, D., Han, J. T., & Lee, C. Y. (2005). Daily
567 consumption of phenolics and total antioxidant capacity from fruit and vegetables in
568 the American diet. *Journal of the Science of Food and Agriculture*, 85(10), 1715-1724,
569 doi:10.1002/jsfa.2176.

570 Dixon, R. A., & Paiva, N. L. (1995). Stress-Induced Phenylpropanoid Metabolism. *The Plant Cell*,
571 7(7), 1085-1097.

572 EC_1441/2007 (2007). Commission regulation on microbiological criteria for foodstuffs. *Official
573 Journal of the European Union*, 32.

574 El Gharras, H. (2009). Polyphenols: food sources, properties and applications – a review.
575 *International Journal of Food Science & Technology*, 44(12), 2512-2518,
576 doi:10.1111/j.1365-2621.2009.02077.x.

577 Gareth Rees, D., & Hugh Jones, D. (1996). Stability of l-phenylalanine ammonia-lyase in
578 aqueous solution and as the solid state in air and organic solvents. *Enzyme and
579 Microbial Technology*, 19(4), 282-288, doi:[http://dx.doi.org/10.1016/0141-
580 0229\(95\)00247-2](http://dx.doi.org/10.1016/0141-0229(95)00247-2).

581 Heredia, J. B., & Cisneros-Zevallos, L. (2009). The effect of exogenous ethylene and methyl
582 jasmonate on pal activity, phenolic profiles and antioxidant capacity of carrots (*Daucus
583 carota*) under different wounding intensities. *Postharvest Biology and Technology*,
584 51(2), 242-249, doi:<http://dx.doi.org/10.1016/j.postharvbio.2008.07.001>.

585 Jacobo-Velázquez, D. A., Martínez-Hernández, G. B., del C. Rodríguez, S., Cao, C.-M., &
586 Cisneros-Zevallos, L. (2011). Plants as Biofactories: Physiological Role of Reactive
587 Oxygen Species on the Accumulation of Phenolic Antioxidants in Carrot Tissue under
588 Wounding and Hyperoxia Stress. *Journal of Agricultural and Food Chemistry*, 59(12),
589 6583-6593, doi:10.1021/jf2006529.

590 Kader, A. A., & Ben-Yehoshua, S. (2000). Effects of superatmospheric oxygen levels on
591 postharvest physiology and quality of fresh fruits and vegetables. *Postharvest Biology
592 and Technology*, 20(1), 1-13, doi:[http://dx.doi.org/10.1016/S0925-5214\(00\)00122-8](http://dx.doi.org/10.1016/S0925-5214(00)00122-8).

593 Kaur, M., & Sharma, H. K. (2013). Effect of enzymatic treatment on carrot cell wall for
594 increased juice yield and effect on physicochemical parameters. *African Journal of*
595 *Plant Science*, 7(6), 234-243.

596 Ke, D., & Saltveit, M. E. (1986). Effects of calcium and auxin on russet spotting and
597 phenylalanine ammonia-lyase activity in iceberg lettuce *HortScience*, 21(5), 1169-1171.

598 Knockaert, G., Lemmens, L., Van Buggenhout, S., Hendrickx, M., & Van Loey, A. (2012).
599 Changes in β -carotene bioaccessibility and concentration during processing of carrot
600 puree. *Food Chemistry*, 133(1), 60-67,
601 doi:<http://dx.doi.org/10.1016/j.foodchem.2011.12.066>.

602 Leistner, L. (2000). Basic aspects of food preservation by hurdle technology. *International*
603 *Journal of Food Microbiology*, 55(1-3), 181-186, doi:[http://dx.doi.org/10.1016/S0168-](http://dx.doi.org/10.1016/S0168-1605(00)00161-6)
604 [1605\(00\)00161-6](http://dx.doi.org/10.1016/S0168-1605(00)00161-6).

605 Lemmens, L., De Vleeschouwer, K., Moelants, K. R. N., Colle, I. J. P., Van Loey, A. M., &
606 Hendrickx, M. E. (2010). β -Carotene Isomerization Kinetics during Thermal Treatments
607 of Carrot Puree. *Journal of Agricultural and Food Chemistry*, 58(11), 6816-6824,
608 doi:10.1021/jf100449t.

609 Martínez-Hernández, G. B., Amodio, M. L., & Colelli, G. (2016). Potential use of microwave
610 treatment on fresh-cut carrots: physical, chemical and microbiological aspects. *Journal*
611 *of the Science of Food and Agriculture*, 96(6), 2063-2072, doi:10.1002/jsfa.7319.

612 Martínez-Hernández, G. B., Artés-Hernández, F., Gómez, P. A., Formica, A. C., & Artés, F.
613 (2013). Combination of electrolysed water, UV-C and superatmospheric O₂ packaging
614 for improving fresh-cut broccoli quality. *Postharvest Biology and Technology*, 76, 125-
615 134, doi:<http://dx.doi.org/10.1016/j.postharvbio.2012.09.013>.

616 Martínez-Hernández, G. B., Gómez, P. A., Pradas, I., Artés, F., & Artés-Hernández, F. (2011).
617 Moderate UV-C pretreatment as a quality enhancement tool in fresh-cut Bimi®
618 broccoli. *Postharvest Biology and Technology*, 62(3), 327-337,
619 doi:<http://dx.doi.org/10.1016/j.postharvbio.2011.06.015>.

620 Marx, M., Stuparic, M., Schieber, A., & Carle, R. (2003). Effects of thermal processing on trans-
621 cis-isomerization of β -carotene in carrot juices and carotene-containing preparations.
622 *Food Chemistry*, 83(4), 609-617, doi:[http://dx.doi.org/10.1016/S0308-8146\(03\)00255-](http://dx.doi.org/10.1016/S0308-8146(03)00255-3)
623 [3](http://dx.doi.org/10.1016/S0308-8146(03)00255-3).

624 Palou, E., López-Malo, A., Barbosa-Cánovas, G. V., Welti-Chanes, J., & Swanson, B. G. (1999).
625 Polyphenoloxidase Activity and Color of Blanched and High Hydrostatic Pressure
626 Treated Banana Puree. *Journal of Food Science*, 64(1), 42-45, doi:10.1111/j.1365-
627 2621.1999.tb09857.x.

628 Patras, A., Brunton, N., Da Pieve, S., Butler, F., & Downey, G. (2009). Effect of thermal and high
629 pressure processing on antioxidant activity and instrumental colour of tomato and
630 carrot purées. *Innovative Food Science & Emerging Technologies*, 10(1), 16-22,
631 doi:<http://dx.doi.org/10.1016/j.ifset.2008.09.008>.

632 Pushkala, R., Parvathy, K. R., & Srividya, N. (2012). Chitosan powder coating, a novel simple
633 technique for enhancement of shelf life quality of carrot shreds stored in macro
634 perforated LDPE packs. *Innovative Food Science & Emerging Technologies*, 16, 11-20,
635 doi:<http://dx.doi.org/10.1016/j.ifset.2012.03.003>.

636 Quitão-Teixeira, L. J., Odriozola-Serrano, I., Soliva-Fortuny, R., Mota-Ramos, A., & Martín-
637 Belloso, O. (2009). Comparative study on antioxidant properties of carrot juice
638 stabilised by high-intensity pulsed electric fields or heat treatments. *Journal of the*
639 *Science of Food and Agriculture*, 89(15), 2636-2642, doi:10.1002/jsfa.3767.

640 Reyes, L. F., Villarreal, J. E., & Cisneros-Zevallos, L. (2007). The increase in antioxidant capacity
641 after wounding depends on the type of fruit or vegetable tissue. *Food Chemistry*,
642 101(3), 1254-1262, doi:<http://dx.doi.org/10.1016/j.foodchem.2006.03.032>.

643 Rodríguez-Verástegui, L. L., Martínez-Hernández, G. B., Castillejo, N., Gómez, P. A., Artés, F., &
644 Artés-Hernández, F. (2015). Bioactive Compounds and Enzymatic Activity of Red

645 Vegetable Smoothies During Storage. [journal article]. *Food and Bioprocess*
646 *Technology*, 9(1), 137-146, doi:10.1007/s11947-015-1609-6.

647 Slavin, J. L., & Lloyd, B. (2012). Health Benefits of Fruits and Vegetables. *Advances in Nutrition:*
648 *An International Review Journal*, 3(4), 506-516, doi:10.3945/an.112.002154.

649 Talcott, S. T., & Howard, L. R. (1999). Phenolic Autoxidation Is Responsible for Color
650 Degradation in Processed Carrot Puree. *Journal of Agricultural and Food Chemistry*,
651 47(5), 2109-2115, doi:10.1021/jf981134n.

652 Tamminen, M., Salminen, S., & Ouwehand, A. C. (2013). Fermentation of carrot juice by
653 probiotics: Viability and preservation of adhesion. *International Journal of*
654 *Biotechnology for Wellness Industries*, 2, 10-15.

655 Tomás-Barberán, F. A., & Espín, J. C. (2001). Phenolic compounds and related enzymes as
656 determinants of quality in fruits and vegetables. *Journal of the Science of Food and*
657 *Agriculture*, 81(9), 853-876, doi:10.1002/jsfa.885.

658 Wen, A., Delaquis, P., Stanich, K., & Toivonen, P. (2003). Antilisterial activity of selected
659 phenolic acids. *Food Microbiology*, 20(3), 305-311,
660 doi:[http://dx.doi.org/10.1016/S0740-0020\(02\)00135-1](http://dx.doi.org/10.1016/S0740-0020(02)00135-1).

661 Yang, Z., Zheng, Y., & Cao, S. (2009). Effect of High Oxygen Atmosphere Storage on Quality,
662 Antioxidant Enzymes, and DPPH-Radical Scavenging Activity of Chinese Bayberry Fruit.
663 *Journal of Agricultural and Food Chemistry*, 57(1), 176-181, doi:10.1021/jf803007j.

664 Yu, L. J., & Rupasinghe, H. P. V. (2012). Effect of acidification on quality and shelf-life of carrot
665 juice. *Canadian Journal of Plant Science*, 92(6), 1113-1120, doi:10.4141/cjps2011-206.

666 Zheng, Y., Wang, C. Y., Wang, S. Y., & Zheng, W. (2003). Effect of High-Oxygen Atmospheres on
667 Blueberry Phenolics, Anthocyanins, and Antioxidant Capacity. *Journal of Agricultural*
668 *and Food Chemistry*, 51(24), 7162-7169, doi:10.1021/jf030440k.

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693 **Figure and table captions**

694 **Figure 1.** Flow diagram of pre-enrichment incubation of carrots, smoothie preparation
695 and storage conditions.

696

697 **Table 1.** pH, titratable acidity (TA; %), soluble solids content (SSC; ° Brix), total color
698 differences (ΔE) and browning index differences (ΔBI) changes of carrot shreds after
699 different treatments (hyperoxia, UVc radiation and controls) during pre-enrichment
700 incubation (72 h at 15 °C) and subsequent storage at 5 °C of prepared non-treated or
701 heat-treated (90 °C for 30 s) smoothie (n=5±SD). Different capital letter denotes
702 significant differences (p < 0.05) among different treatments for the same sampling day.
703 Different lowercase letter denotes significant differences (p < 0.05) among different
704 sampling days for the same treatment.

705

706 **Table 2.** Mesophilic, psychrophilic and yeasts and molds counts (log CFU g⁻¹) of carrot
707 shreds after different treatments (hyperoxia, UVc radiation and controls) during pre-
708 enrichment incubation (72 h at 15 °C) and subsequent storage at 5 °C of prepared non-
709 treated or heat-treated (90 °C for 30 s) smoothie (n=5±SD). (n=5±SD). Different capital
710 letter denotes significant differences (p < 0.05) among different treatments for the same
711 sampling day. Different lowercase letter denotes significant differences (p < 0.05)
712 among different sampling days for the same treatment.

713

714 **Table 3.** Phenylalanine ammonia-lyase activity (PAL; µmol *t*-cinnamic acid synthesized
715 kg⁻¹ fw h⁻¹), total phenolic content (TPC; Chlorogenic acid equivalent kg⁻¹ fw) and total
716 antioxidant capacity (TAC; mg Trolox equivalents kg⁻¹ fw) of carrot shreds after
717 different treatments (hyperoxia, UVc radiation and controls) during pre-enrichment
718 incubation (72 h at 15 °C) and subsequent storage at 5 °C of prepared non-treated or
719 heat-treated (90 °C for 30 s) smoothie (n=5±SD). (n=5±SD). Different capital letter
720 denotes significant differences (p < 0.05) among different treatments for the same
721 sampling day. Different lowercase letter denotes significant differences (p < 0.05)
722 among different sampling days for the same treatment.

723

724 **Table 4.** Individual phenolic contents (mg kg⁻¹ fw) of carrot shreds after different
725 treatments (hyperoxia, UVc radiation and controls) during pre-enrichment incubation
726 (72 h at 15 °C) and subsequent storage at 5 °C of prepared non-treated or heat-treated
727 (90 °C for 30 s) smoothie (n=5±SD). (n=5±SD). Different capital letter denotes
728 significant differences (p < 0.05) among different treatments for the same sampling day.
729 Different lowercase letter denotes significant differences (p < 0.05) among different
730 sampling days for the same treatment.

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